

HYPOIODITE REACTION OF N-ACETYLJERVINE:
THE FORMATIONS OF OXEPANE AND OXETANE RINGS BY HETERO ATOM INSERTION
FROM ALKOXYL RADICAL (1)

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In a preceding paper (1) we clarified the structure of a novel dimeric compound 13, one of the five products formed in the hypiodite reaction of N-acetyljervine 1. (2) In this paper we discuss the structures of another four products, 7, 12, 14 and 16, obtained in this reaction and the probable sequences of the formation of these products including the dimeric ether 13 discussed in the preceding paper. (3)

As in the case of the dimer 13 (1) it was apparent from the i.r., u.v., and ¹H n.m.r. spectra that the C, D, and the heterocyclic rings of all four products were unaltered. (4) The structure of 7, m.p. 159-160°, (5% yield) which contained iodine, was proved to be N-acetyl-4-iodo-11-oxo-3,4-seco-jerv-5,12(13)-dien-3-al on the following ground. (5)

In the i.r. spectrum, except a series of bands due to the N-Ac, α,β -unsaturated carbonyl, no band due to the hydroxyl group was observed. In the n.m.r. spectrum a one-proton singlet at τ 1.91, a one-proton doublet at τ 4.06 ($J=4.5$ Hz) and a two-proton triplet at τ 5.40 ($J=15.0$ Hz) were observed and these signals are safely assignable to the C-3 formyl proton, an olefinic proton at the C-6 and the allylic C-4 methylene protons (6) flanked further by an iodine. The n.m.r. and i.r. spectra of 12, m.p. 218-220° (2% yield), led us to assign N-acetyl-11-oxo-oxidomethylene-3 α ,5 α -A-nor-jerv-5,12(13)-diene as the structure. In the i.r. spectrum, no hydroxyl band was observed. The n.m.r. spectrum showed a broad one-proton singlet at τ 4.48 ($W_H=3.9$ Hz), a one-proton quartet at τ 5.55 ($J_{AX}^{ap}=4.5$, $J_{BX}^{ap}=12.3$) and a two-proton double doublet of AB system

at τ 5.84 and τ 6.25 (J_{AB} =7.6 Hz). These signals are assignable to the C-3 β -H and the C-6 β -H attached to the iodine-bearing carbon and isolated methylene protons incorporated in the oxetane ring in a way entirely analogous to the case of oxidomethylene3 α ,5 α -A-nor-cholestane. (7)

We assign N-acetyl-3,11-dioxo-4-oxa-A-homo-jerv-5,12(13)-diene structure for the product 16, m.p. 258-260° (1% yield) (mass, m/e 481, M⁺). In the i.r. spectrum of 16 again no OH band was observed, but as well as bands due to the α , β -unsaturated carbonyl (1640 and 1714 cm⁻¹), a new unstrained lactonic carbonyl band (1737 cm⁻¹) was present. In the n.m.r. there were observable typical double doublets attributable to AB system at τ 5.84 and τ 5.36 (J_{AB} =13.5 Hz) and a one-proton doublet of olefinic proton at τ 4.21 (J=3.0 Hz). These protons are assignable to the C-4a methylene protons of the oxepane ring and to the C-6-H.

Finally, the structure of 14 m.p. 165-167°, (<1%) (mass, m/e 453, M⁺-14) was clarified as N-acetyl-11-oxo-4-oxa-A-homo-jerv-5,12(13)-diene as follows: in the i.r. spectrum there was no band due to the hydroxyl group. In the n.m.r. spectrum, there was an olefinic one-proton signal at τ 4.57 (J=3.0 Hz) and a nine-proton multiplet from τ 5.81 to τ 7.20. The splitting pattern of the signals of the multiplet was nearly the same as the superimposed signals of the proton signals of C-22-H, C-23-H, C-27-methylene and C-9-H of N-acetyljervine and those of C-3-methylene and C-4a methylene of 4-oxa-A-homo-cholest-5-ene. (7).

The probable sequences of the formations of the five compounds 7, 12, 14, 16 and 13 are summarized in the Scheme. The common intermediate for all five products is considered to be the allyl radical 5. For the formation of oxetane 12 there can be two probable pathways A and B as we previously proposed (7), and on the basis of the same stereoelectronic considerations as two of the present authors suggested (7), the formation of 12 via the Paterno-Büchi reaction of the preformed olefinic aldehyde 8 could be excluded and the formation of 12 via pathway B is more probable. This pathway is considered to involve either successive radical cyclization to form more stable 3 α ,5 α -oxetane (e.g., 5 \rightarrow 9 \rightarrow 12), or the formation of oxetane via attack of electrophilic oxygen of the excited 5 on the allyl radical moiety. On the other hand, in order to explain the formations of several compounds with oxepane rings, we must assume another radical

intermediate, 9, from which ramification will start to afford 14 by the hydrogen abstraction from the solvent, to give 16 via the formation of jodohydrin 11 followed by oxidation, and to 13 by combination with alkoxy radical 3.

We have already assumed the intervention of the intermediate such as 9 in the oxetane formation in the hypiodite reaction of normal steroids. (6) The formation of several oxepane derivatives such as 13, 14 and 16 in the present case may support our original proposal, i.e., the formation of oxetanes (e.g., 12) via successive radical cyclization from the allyl radical (e.g., 5).

It is rather remarkable that while the products of the hypiodite reaction in the steroid series were merely 3,4-seco-aldehyde (e.g., 7) and oxetanes (e.g., 12), several oxepane derivatives were obtained in the hypiodite reaction of C-nor-D-homosteroid.

It would not be wise to attribute this clear variation of products between steroids and C-nor-D-homosteroids to any single effect. However, a comparison of Dreiding models of assumed intermediates 9 derived from steroids and C-nor-D-homosteroids indicates that the modification of the C-ring of 9 may introduce a subtle difference in the predominant conformation of the flexible oxepane ring of 9, and this might be one of the important factors leading to the intermolecular radical addition to lead to the types 13, 14 and 16 over the intramolecular radical cyclization to oxetanes in the C-nor-D-homosteroid series.

REFERENCES AND NOTES

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- (3) See for review of hypiodite reaction. K. Heusler and J. Kalvoda, Angew. Chem., 76, 518 (1964); J. Kalvoda and K. Heusler, Synthesis, 501 (1971).
- (4) Details of these spectral data will be published in our full paper. All the spectra in this paper were taken in CDCl₃ by 100 MHz instrument.
- (5) In this paper, we used "jervane" nomenclature for the C₂₇ compounds. Cf., F. C. Chang and R. C. Ebersole, Tetrahedron Letters, 3521 (1968).
- (6) Nonequivalence of these protons are due to the restriction of rotation of C-4-C-5 bond.
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